

PHASE BEHAVIOUR STUDY OF GAS HYDRATES: POTENTIAL OPPORTUNITIES
FOR CO₂ SEPARATION

by

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ABSTRACT

The presence of high CO₂ in certain natural gas reservoirs has posted huge challenges in the natural gas purification. Although there are several separation technologies like amine absorption and cryogenic process, their separation capability is limited to low CO₂ in natural gas. An alternative approach to separate high CO₂ from CO₂-CH₄ mixtures by formation of CO₂ hydrate is being explored in this work. The objectives were to study the phase behaviour of various gas hydrates and to investigate the effect of pressure, temperature, water and CO₂ compositions towards the separation of CO₂ from CO₂-CH₄ mixtures via formation of CO₂ hydrates. The research work was carried out by using an established thermodynamics hydrate program, CSMGem. This work was successfully predicted the equilibrium pressures of hydrate-liquid water-vapor (H-L_w-V) line for single hydrates (CH₄, C₂H₆, C₃H₈, CO₂ and N₂) with calculated AAD% less than 5%. The formation of single hydrates was found to be more favourable by increasing operating pressure and decreasing temperature. The AAD% was further improved by introducing pressure correction factor known as cavity constant, λ . It was found that the calculated AAD% for CO₂ and C₃H₈ hydrates were improved by 20% and 27%, respectively. The formations of binary hydrates (CH₄-C₂H₆, CH₄-C₃H₈, CH₄-CO₂ and CH₄-N₂) were found to depend on the type and compositions of gases present. The equilibrium lines of mixed hydrates were located in between those of single hydrates. Introduction of cavity constant of mixture, λ_{mix} has generally improved the predicted data for all binary hydrates. The highest improvement obtained was 69% for CH₄-C₃H₈ hydrate with CH₄ composition of 36.2%. The separation factor on separation of CO₂ from CO₂-CH₄ mixture was found in the range of 1.0 to 2.4 in the operating conditions studied. The separation factor was increasing with decreasing of pressures, temperatures and CO₂ compositions and vice-versa with water compositions. Introduction of multistage separation has further improved the separation factor from 2.3 to 22.5 for gas mixture with 30% of CO₂. However, it is not recommended for high CO₂ concentration as the number of moles in vapour phase at each stage decreases with increasing number of stages. It can be

concluded that the formation of CO₂ hydrate has a great potential to be further explored since it can be used to address separation of CO₂ from CO₂-CH₄ mixtures.