Removal of Para-nitrophenol from Industrial Waste Water (IWW) by Adsorption using Modified Activated Carbon

by

Goh Hui San

Dissertation submitted in partial fulfillment of the requirements of the Bachelor of Engineering (Hons) (Chemical Engineering)

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Universiti Teknologi PETRONAS Bandar Seri Iskandar 31750 Tronoh Perak Darul Ridzuan

CERTIFICATION OF APPROVAL

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A project dissertation submitted to the Chemical Engineering Programme Universiti Teknologi PETRONAS in partial fulfillment of the requirement for the BACHELOR OF ENGINEERING (Hons) (CHEMICAL ENGINEERING)

Approved by,

(Dr. Usama Mohamed Nour El Demerdash)

UNIVERSITI TEKNOLOGI PETRONAS TRONOH, PERAK May 2011

CERTIFICATION OF ORIGINALITY

This is to certify that I am responsible for the work submitted in this project, that the original work is my own except as specified in the references and acknowledgements, and that the original work contained herein have not been undertaken or done by unspecified sources or persons.

GOH HUI SAN

ABSTRACT

Para-nitrophenol is one of the organic pollutants found in industrial wastewater treatment. Its toxicity can bring great harm to biological systems, including plants, fish and other organisms. Besides, para-nitrophenol will defect resins when entering the ion exchange unit in Condensate Polishing Plant (one part of the wastewater treatment system). Therefore, activated carbon used in the carbon filter installed before the ion exchange unit has to be modified to overcome the resin life problems as a result of fouling due to the organic compounds by adsorbing the organic pollutants more efficiently. For this research, surface modification of activated carbon using acidic solution is applied. The effect of adsorbate concentration, pH and temperature on the adsorption capacity for both unmodified and modified activated carbon are observed as well as analyzed. Scanning Electron Microscope (SEM) is used to study the pore structure of the activated carbon before and after modification, whereas UV-vis Spectrophotometer is used to determine the adsorption capacity of the activated carbon towards organic pollutant. As the concentration of absorbate increases, adsorption capacity increases. The adsorption reaches its highest point around pH 7. When the carbonization temperature increases, the adsorption capacity increases as well.

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TABLE OF CONTENTS

ABST	RACT	iii
ACKI	NOWLEDGEMENTS	iv
LIST	OF FIGURES	vii
LIST	OF TABLES	viii
ABBF	REVIATIONS AND NOMENCLATURES	ix
CHAI	PTER 1	Error! Bookmark not defined.
INT	RODUCTION	
1.1	Background	Error! Bookmark not defined.
1.2	Problem Statement	
1.3	Objectives	
1.4	Scope of Study	
CHAI	PTER 2	
LIT	ERATURE REVIEW AND THEORY	
2.1	Organic Pollution in Wastewater	5
2.2	Adsorption	
	2.2.1 Adsorption Isotherm	7
	2.2.2 Activated Carbon	9
2.3	Surface Modifications of Activated Carbon	n.Error! Bookmark not defined.
	2.3.1 Chemical Modification	
	2.3.2 Physical Modification	
	2.3.3 Biological Modification	
2.4	Para-nitrophenol	Error! Bookmark not defined.
2.5	Condensate Polishing Plant	Error! Bookmark not defined.
CHAI	PTER 3	Error! Bookmark not defined.
ME'	THODOLOGY	Error! Bookmark not defined.
3.1	Research Methodology	Error! Bookmark not defined.
	3.1.1 Preparation of Activated Carbon	16
	3.1.2 Chemical Treatment of Activated Carb	oon17
	3.1.3 Carbonization of Activated Carbon	
	3.1.4 Adsorption Study	19
	3.1.5 Characterization	

3.2	Project Activities	
3.3	Gantt Chart	
3.4	Tools Required	Error! Bookmark not defined.

CHAPTER 4Error! Bookmark not		
RES	SULTS AND DISCUSSSIONError! Bookm	ark not defined.
4.1	Preparation of Activated CarbonError! Bookm	ark not defined.
4.2	Chemical Treatment of Activated CarbonError! Bookm	ark not defined.
4.3	Carbonization of Activated CarbonError! Bookm	ark not defined.
4.4	Characterization	
	4.4.1 Scanning Electron Microscope (SEM)	
	4.4.2 Fourier Transform Infrared Spectroscopy (FTIR)	
4.5	Adsorption Study	
	4.5.1 Calibration Curve	
	4.5.2 Equilibrium Studies	
	4.5.3 Effect of Adsorbate Concentration on Adsorption Capa	city36
	4.5.4 Effect of pH on Adsorption Capacity	
	4.5.5 Effect of Adsorbate Concentration on Adsorption Capa	city38
CHAP	PTER 5	
CON	NCLUSION AND RECOMMENDATION	
5.1	Conclusion	
5.2	Recommendation	
REFE	ERENCES	
APPE	ENDICES	

LIST OF FIGURES

Figure 1: Definition sketch for adsorption of an organic constituent	6
Figure 2: Langmuir isotherm curve	8
Figure 3: Freundlich isotherm curve	9
Figure 4: Modification techniques for AC	11
Figure 5: Condensate polishing plant	15
Figure 6: UV-vis Spectrophotometer	19
Figure 7: Scanning Electron Microscope (SEM)	20
Figure 8: Fourier Transform Infrared Spectroscopy (FTIR)	21
Figure 9: Raw activated carbon	25
Figure 10: Clean activated carbon	25
Figure 11: Oxidized activated carbon	25
Figure 12: Oven	26
Figure 13: Turbular furnace	26
Figure 14a: Raw AC at 500X magnification	28
Figure 14b: Raw AC at 5000X magnification	29
Figure 15a: Oxidized AC at 500X magnification	29
Figure 15b: Oxidized AC at 5000X magnification	
Figure 16a: Carbonized AC at 500X magnification	30
Figure 16b: Carbonized AC at 5000X magnification	31
Figure 17: FTIR spectra for GAC, CAC and MAC	32
Figure 18: Wavelength for calibration data	33
Figure 19: Calibration curve	34
Figure 20: Equilibrium studies	35
Figure 21: Q _e for different concentration	36
Figure 22: Q _e for different pH	
Figure 23: Q _e for different temperature	39

LIST OF TABLES

Table 1: List of chemicals	24
Table 2: List of equipment.	24
Table 3: Experimental data for calibration curve	33
Table 4: Experimental data for equilibrium studies	35
Table 5: Data for different adsorbate concentration for GAC	36
Table 6: Data for different adsorbate concentration for MAC	36
Table 7: Data for different pH for GAC	37
Table 8: Data for different pH for MAC	37
Table 9: Data for different temperature for GAC	38
Table 10: Data for different temperature for MAC	39

ABBREVIATIONS AND NOMENCLATURES

AC	Activated Carbon
CAC	Clean Activated Carbon
FTIR	Fourier Transform Infrared Spectroscopy
GAC	Granular Activated Carbon
HCl	Hydrochloric Acid
HNO ₃	Nitric Acid
MAC	Modified Activated Carbon
NaOH	Sodium Hydroxide
SEM	Scanning Electron Microscope
US EPA	United States Environmental Protection Agency
VOC	Volatile Organic Compound